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1,2-Bis{4-[1-(anthracen-9-ylmethyl)-1*H*-1,2,3-triazol-4-yl]phenyl}-1,2-bis[4,5-bis(methylsulfanyl)-1,3-dithiol-2-ylidene]ethane

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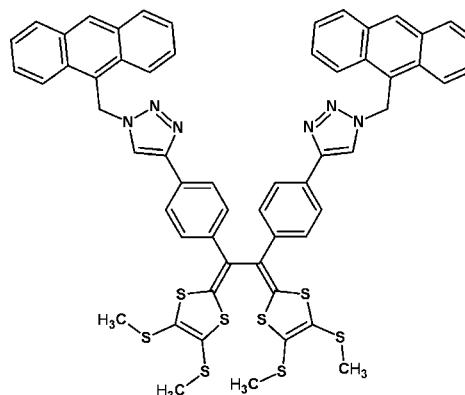
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Key indicators: single-crystal X-ray study; $T = 168$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; disorder in main residue; R factor = 0.058; wR factor = 0.160; data-to-parameter ratio = 14.9.

The title molecule, $\text{C}_{58}\text{H}_{44}\text{N}_6\text{S}_8$, has point symmetry 2 (in the Schönflies notation C_2). The related crystallographic twofold axis bisects the central ethane bond while it is parallel to the monoclinic unique axis of the unit cell. The dithiole=C—C=dithiole torsion angle is $103.7(4)^\circ$ and the triazole–anthracene moieties adopt a pincer-like conformation. The crystal structure features C—H \cdots S and C—H \cdots N contacts. The distance between the stacked anthracene fragments [centroid–centroid separations of $3.6871(19)$ Å, off-set by $1.516(3)$ Å and mean anthracene plane–plane separations of $3.361(2)$ Å], which are parallel to (101) and $(\bar{1}01)$, indicates intermolecular anthracene–anthracene π – π contacts. One of the terminal methylsulfanyl groups was modelled as being disordered with two refined orientations that converged to occupancies of 0.809 (5) and 0.191 (5).

Related literature

The simpler analogues, 4,4',5,5'-tetramethylthiolato-2,2'-ethanediyldene(1,3-dithiole) and bis(4,5-bis(methylthio)-1,3-dithiol-2-ylidene)succinonitrile, were previously reported by Bryce *et al.* (1996) and Jia *et al.* (2005), respectively. For information on the photophysical properties of tetrathiafulvalene vinyllogues, see: Mulla *et al.* (2012). For synthesis *via* Cu-catalysed alkyne–azide coupling reactions employed for the title compound, see: Meldal & Tornøe (2008); Hein *et al.* (2010); Zhao *et al.* (2012). For a discussion of hydrogen bonding, including non-traditional interactions, see: Desiraju (2011); Arunan *et al.* (2011). For standard bond lengths, see: Allen *et al.* (1987) and for a description of the Cambridge Structural Database, see Allen (2002).



Experimental

Crystal data

 $\text{C}_{58}\text{H}_{44}\text{N}_6\text{S}_8$ $M_r = 1081.47$ Monoclinic, $C2/c$ $a = 15.906(4)$ Å $b = 14.737(3)$ Å $c = 21.570(5)$ Å $\beta = 98.846(7)^\circ$ $V = 4996(2)$ Å³ $Z = 4$ Mo $K\alpha$ radiation $\mu = 0.41$ mm⁻¹ $T = 168$ K $0.27 \times 0.23 \times 0.12$ mm

Data collection

Rigaku AFC8 diffractometer

Absorption correction: multi-scan

(REQAB; Jacobson, 1998)

 $T_{\min} = 0.765$, $T_{\max} = 0.952$

24431 measured reflections

5169 independent reflections

4127 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.059$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.058$ $wR(F^2) = 0.160$ $S = 1.06$

5169 reflections

347 parameters

3 restraints

H-atom parameters constrained

 $\Delta\rho_{\max} = 0.70$ e Å⁻³ $\Delta\rho_{\min} = -0.62$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C28}-\text{H28C}\cdots\text{N2}^{\text{i}}$	0.98	2.55	3.495 (5)	161
$\text{C25}-\text{H25}\cdots\text{S2}^{\text{ii}}$	0.95	3.00	3.784 (4)	141

Symmetry codes: (i) $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; (ii) $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$.

Data collection: *CrystalClear-SM Expert* (Rigaku, 2009); cell refinement: *CrystalClear-SM Expert*; data reduction: *CrystalClear-SM Expert*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *OLEX2* (Dolomanov *et al.*, 2009), *publCIF* (Westrip, 2010) and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FB2271).

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supplementary materials

Acta Cryst. (2012). E68, o3298–o3299 [doi:10.1107/S1600536812045254]

1,2-Bis{4-[1-(anthracen-9-ylmethyl)-1*H*-1,2,3-triazol-4-yl]phenyl}-1,2-bis[4,5-bis(methylsulfanyl)-1,3-dithiol-2-ylidene]ethane

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Comment

The asymmetric unit of the title molecule contains a half of the molecule. The whole molecule has the point symmetry 2 (or C_2 in the Schönflies notation). The twofold axis is parallel to the monoclinic axis and passes through the centre of the C4—C4ⁱ bond ($i = -x, y, 1/2 - z$; Fig. 1). This point symmetry has been previously reported for the related molecule, bis-(4,5-bis(methylthio)-1,3-dithiol-2-ylidene)succinonitrile (Jia *et al.* 2005). The dithiole=C—C=dithiole torsion angle is 103.7 (4)° for the title molecule, while for the related analogue Jia *et al.* (2005) reported 91.7° (no reported s.u.). In the Cambridge Crystallographic Database (Allen, 2002), the article by Jia *et al.* (2005) is deposited as CCDC 285669, refcode JAVWEJ, and the pertinent torsion angle has been given as 85.9 (7)°. The larger value of this torsion angle in the title molecule may result from the adoption of a pincer-like conformation of the triazole and anthracene fragments. Both the title molecule and that reported by Jia *et al.* (2005) exhibit a twisted *cis* conformation about the C4—C4ⁱ bond ($i = -x, y, 1/2 - z$), which contrasts sharply with the planar 2,2'-ethanediylidene(1,3-dithiole) framework reported for 4,4',5,5'-tetramethylthiolato-2,2'-ethanediylidene(1,3-dithiole) (Bryce *et al.*, 1996), which adopts a *trans* conformation.

One of the terminal S—CH₃ groups was disordered with two refined orientations that converged to occupancy 0.809 (5) for S4, C29 and the attached methyl H atoms H29A, H29B, H29C, and 0.191 (5) for S4A, C29A and the attached methyl H atoms H29D, H29E, H29F. (These hydrogens were difficult to discern in the difference map.)

Non-traditional D—H⋯A interactions (Arunan *et al.*, 2011, and Desiraju, 2011) are present between C25_{anthracene}—H25⋯S2ⁱⁱ ((ii) = $1/2 - x, 3/2 - y, 1 - z$) and C28_{methyl}—H28C⋯N2ⁱⁱⁱ ((iii) = $x + 1/2, -y + 3/2, z + 1/2$, see Fig. 2 and Table 1). Further intermolecular interactions in the form of π -electron_{anthracene}— π -electron_{anthracene} contacts with centroid—centroid separations of 3.6871 (19) Å, off-set by 1.516 (3) Å and mean anthracene plane-plane separations of 3.361 (2) Å are present (Fig. 3; as measured from the anthracene fragment in x, y, z to (iv) = $1/2 - x, 5/2 - y, 1 - z$). The molecules that exhibit H-bonding and π -electron— π -electron interactions are parallel to the (1, 0, 1) and (-1, 0, 1) planes (determined using *OLEX2* by Dolomanov *et al.*, 2009.)

Experimental

The title compound was synthesized by the procedures reported by Mulla *et al.* (2012). 3 mg of the title compound were added to 0.8 ml of chloroform in a vial and heated to 323 K to give a clear solution. This solution was first cooled to room temperature and then methanol (0.5 ml) was added slowly to the chloroform solution by diffusion at 277 K over 3 days. After this time, yellow prismatic crystals with dimensions similar to those of the measured sample (0.27 × 0.23 × 0.12 mm), formed.

Refinement

While all hydrogen atoms appeared in the difference electron density map, except those pertinent to the disordered groups, they were introduced into idealized positions and refined using the riding atom formalism. The applied constraints were: $C_{\text{aryl}}\text{---}H_{\text{aryl}} = 0.95 \text{ \AA}$, $C_{\text{methyl}}\text{---}H_{\text{methyl}} = 0.98 \text{ \AA}$, $C_{\text{methylene}}\text{---}H_{\text{methylene}} = 0.99 \text{ \AA}$; $U_{\text{iso}}(H_{\text{aryl}}) = 1.2U_{\text{eq}}(C_{\text{aryl}})$, $U_{\text{iso}}(H_{\text{methylene}}) = 1.2U_{\text{eq}}(C_{\text{methylene}})$, $U_{\text{iso}}(H_{\text{methyl}}) = 1.5U_{\text{eq}}(C_{\text{methyl}})$. One terminal S—CH₃ group was disordered with two orientations for S4—C29 (and the pertinent H-atoms) and S4A—C29A (and the pertinent H-atoms) the occupancies of which were constrained to equal to 1. (The respective occupancies resulted in 0.809 (5) and 0.191 (5).) A distance restraint (the command SADI from *SHELXL97* by Sheldrick, 2008) was applied to the S3—C28, S4—C29 and S4A—C29A bonds. The converged S—C bond lengths were consistent with those reported by Allen *et al.* (1987).

Computing details

Data collection: *CrystalClear-SM Expert* (Rigaku, 2009); cell refinement: *CrystalClear-SM Expert* (Rigaku, 2009); data reduction: *CrystalClear-SM Expert* (Rigaku, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *OLEX2* (Dolomanov *et al.*, 2009), *publCIF* (Westrip, 2010) and *PLATON* (Spek, 2009).

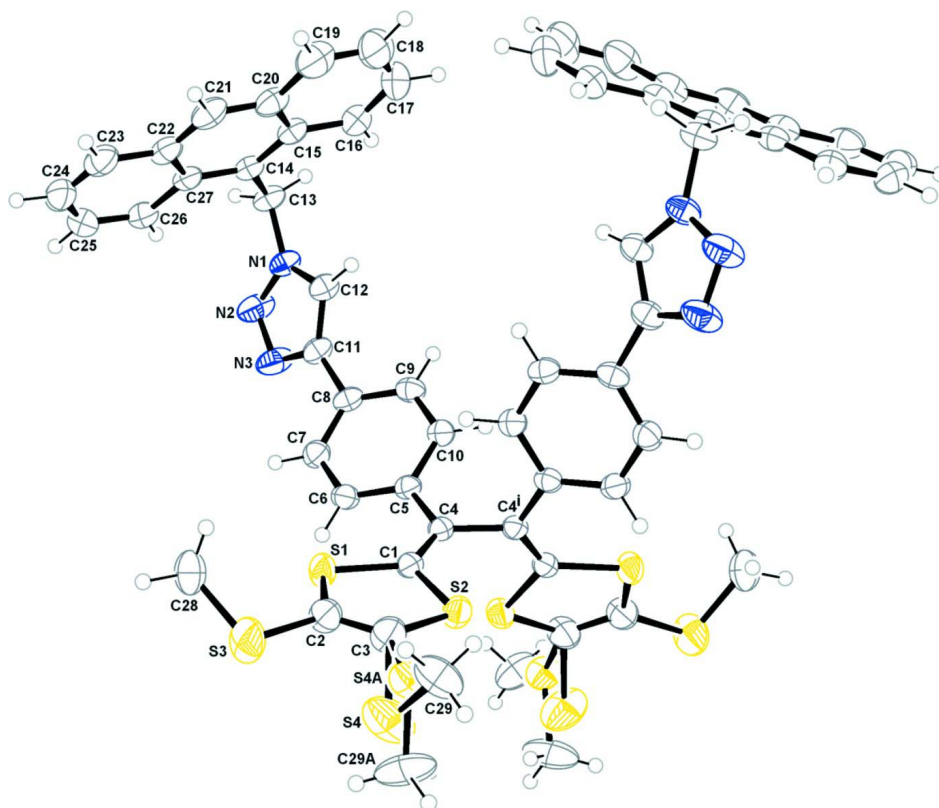
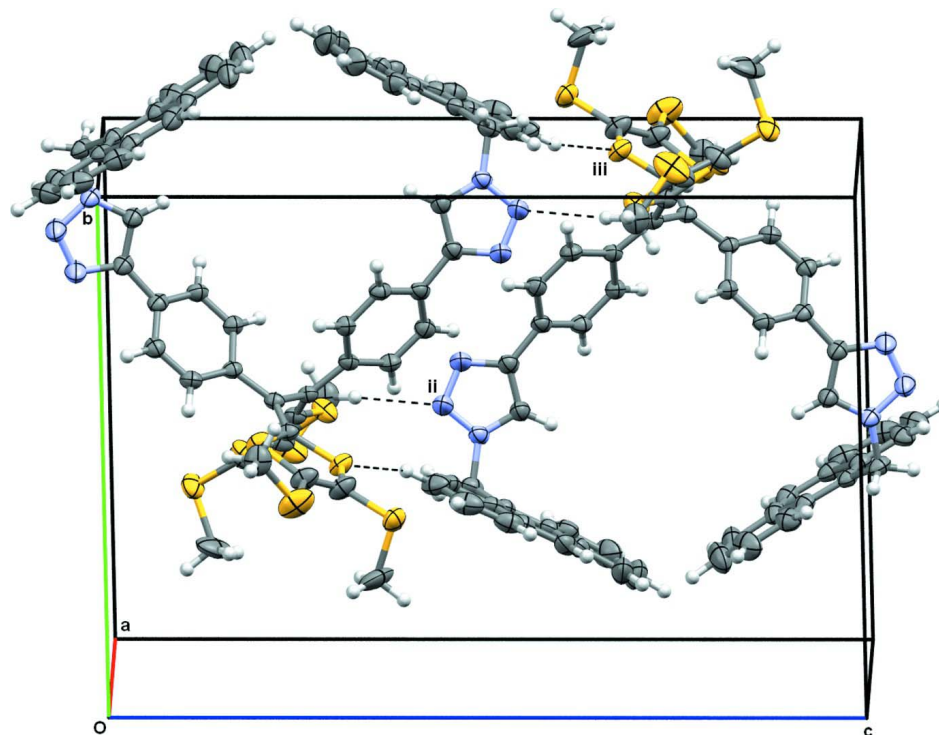
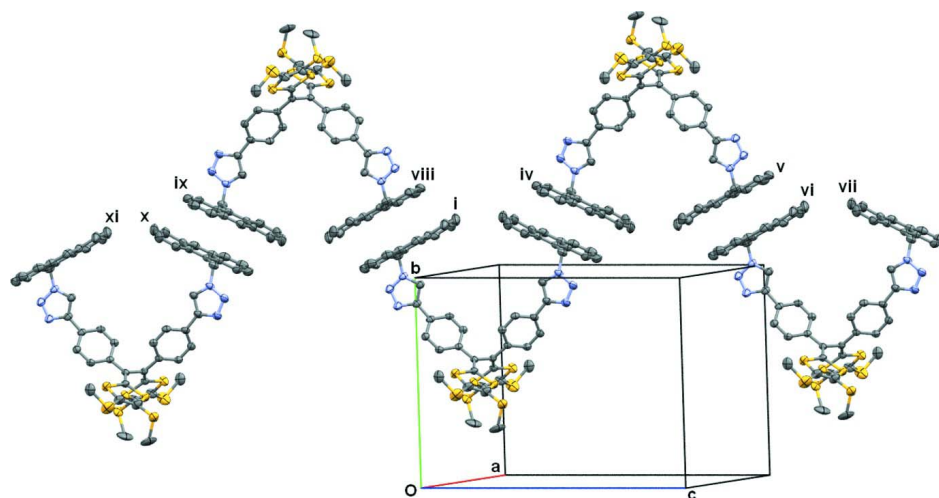


Figure 1

View of the title molecule. The displacement ellipsoids are shown at the 50% probability level. Symmetry code: (i) = $-x, y, 1/2 - z$.


Figure 2

Short C—H⋯S and C—H⋯N contacts between the molecules of the title compound. Symmetry codes: (i) $-x, y, 1/2 - z$; (ii) $1/2 - x, 3/2 - y, 1 - z$; (iii) $1/2 + x, 3/2 - y, 1/2 + z$.


Figure 3

π -electron— π -electron interactions between the anthracene units, propagating parallel to $(1, 0, 1)$, with the centroid-centroid separations of $3.6871(19)$ Å (as measured between x, y, z and (iv) $= 1/2 - x, 5/2 - y, 1 - z$). Symmetry codes: (i) $-x, y, 1/2 - z$; (iv) $1/2 - x, 5/2 - y, 1 - z$; (v) $1/2 + x, 5/2 - y, 1/2 + z$; (vi) $1 - x, y, 3/2 - z$; (vii) $1 + x, y, 1 + z$; (viii) $-1/2 + x, 5/2 - y, -1/2 + z$; (ix) $-1/2 - x, 5/2 - y, -z$; (x) $-1 + x, y, -1 + z$; (xi) $-1 - x, y, -1/2 - z$.

1,2-Bis{4-[1-(anthracen-9-ylmethyl)-1H-1,2,3-triazol-4-yl]phenyl}-1,2-bis[4,5-bis(methylsulfanyl)-1,3-dithiol-2-ylidene]ethane

Crystal data

$C_{58}H_{44}N_6S_8$	$F(000) = 2248$
$M_r = 1081.47$	$D_x = 1.438 \text{ Mg m}^{-3}$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation, $\lambda = 0.71075 \text{ \AA}$
Hall symbol: $-C 2yc$	Cell parameters from 12368 reflections
$a = 15.906 (4) \text{ \AA}$	$\theta = 3.3\text{--}27.6^\circ$
$b = 14.737 (3) \text{ \AA}$	$\mu = 0.41 \text{ mm}^{-1}$
$c = 21.570 (5) \text{ \AA}$	$T = 168 \text{ K}$
$\beta = 98.846 (7)^\circ$	Prism, yellow
$V = 4996 (2) \text{ \AA}^3$	$0.27 \times 0.23 \times 0.12 \text{ mm}$
$Z = 4$	

Data collection

Rigaku AFC8 diffractometer	24431 measured reflections
Radiation source: fine-focus sealed tube	5169 independent reflections
Graphite - Rigaku SHINE monochromator	4127 reflections with $I > 2\sigma(I)$
Detector resolution: $14.63 \text{ pixels mm}^{-1}$	$R_{\text{int}} = 0.059$
ω scans	$\theta_{\text{max}} = 26.5^\circ$, $\theta_{\text{min}} = 3.3^\circ$
Absorption correction: multi-scan (<i>REQAB</i> ; Jacobson, 1998)	$h = -19 \rightarrow 19$
$T_{\text{min}} = 0.765$, $T_{\text{max}} = 0.952$	$k = -18 \rightarrow 18$
	$l = -27 \rightarrow 24$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.058$	Hydrogen site location: difference Fourier map
$wR(F^2) = 0.160$	H-atom parameters constrained
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.0767P)^2 + 8.5759P]$
5169 reflections	where $P = (F_o^2 + 2F_c^2)/3$
347 parameters	$(\Delta/\sigma)_{\text{max}} < 0.001$
3 restraints	$\Delta\rho_{\text{max}} = 0.70 \text{ e \AA}^{-3}$
97 constraints	$\Delta\rho_{\text{min}} = -0.62 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.21077 (4)	0.55797 (5)	0.29159 (4)	0.0438 (2)	
S2	0.08541 (4)	0.49862 (5)	0.18534 (4)	0.0415 (2)	

S3	0.35601 (5)	0.46517 (8)	0.24249 (6)	0.0746 (3)	
S4	0.21209 (7)	0.41217 (8)	0.11540 (6)	0.0497 (5)	0.809 (5)
C29	0.2002 (4)	0.2963 (3)	0.1414 (3)	0.0846 (19)	0.809 (5)
H29A	0.2504	0.2791	0.1714	0.127*	0.809 (5)
H29B	0.1943	0.2554	0.1052	0.127*	0.809 (5)
H29C	0.1493	0.2919	0.1618	0.127*	0.809 (5)
S4A	0.2162 (5)	0.3602 (6)	0.1509 (3)	0.086 (3)	0.191 (5)
C29A	0.1999 (15)	0.4153 (16)	0.0740 (10)	0.073 (6)	0.191 (5)
H29D	0.1510	0.4564	0.0711	0.109*	0.191 (5)
H29E	0.1891	0.3691	0.0411	0.109*	0.191 (5)
H29F	0.2508	0.4500	0.0687	0.109*	0.191 (5)
N1	0.11476 (15)	1.01534 (15)	0.50733 (10)	0.0358 (5)	
N2	0.1323 (2)	0.95560 (17)	0.55383 (12)	0.0527 (7)	
N3	0.1189 (2)	0.87389 (17)	0.52924 (12)	0.0538 (7)	
C1	0.10424 (15)	0.56405 (17)	0.25385 (12)	0.0318 (5)	
C2	0.24892 (19)	0.4963 (2)	0.23200 (16)	0.0487 (8)	
C3	0.1915 (2)	0.4702 (2)	0.18276 (17)	0.0513 (8)	
C4	0.04180 (15)	0.61453 (16)	0.27167 (12)	0.0289 (5)	
C5	0.05131 (15)	0.67558 (16)	0.32632 (12)	0.0297 (5)	
C6	0.11164 (16)	0.66476 (17)	0.38029 (12)	0.0343 (6)	
H6	0.1458	0.6115	0.3846	0.041*	
C7	0.12316 (17)	0.72875 (18)	0.42735 (13)	0.0366 (6)	
H7	0.1642	0.7188	0.4636	0.044*	
C8	0.07492 (17)	0.80792 (17)	0.42201 (12)	0.0339 (6)	
C9	0.01149 (17)	0.81784 (17)	0.37050 (13)	0.0346 (6)	
H9	-0.0235	0.8704	0.3671	0.041*	
C10	-0.00144 (16)	0.75248 (17)	0.32412 (12)	0.0319 (5)	
H10	-0.0467	0.7596	0.2902	0.038*	
C11	0.09216 (17)	0.88302 (18)	0.46645 (12)	0.0360 (6)	
C12	0.08963 (17)	0.97300 (18)	0.45269 (13)	0.0371 (6)	
H12	0.0734	1.0002	0.4127	0.045*	
C13	0.13241 (18)	1.11193 (18)	0.51750 (14)	0.0390 (6)	
H13A	0.1583	1.1224	0.5617	0.047*	
H13B	0.0787	1.1468	0.5092	0.047*	
C14	0.19234 (17)	1.14373 (17)	0.47421 (12)	0.0336 (6)	
C15	0.16101 (18)	1.18757 (18)	0.41722 (13)	0.0383 (6)	
C16	0.0730 (2)	1.2060 (2)	0.39650 (16)	0.0506 (8)	
H16	0.0316	1.1858	0.4207	0.061*	
C17	0.0483 (3)	1.2525 (3)	0.34189 (19)	0.0704 (11)	
H17	-0.0105	1.2639	0.3289	0.085*	
C18	0.1066 (3)	1.2840 (3)	0.30463 (17)	0.0723 (11)	
H18	0.0878	1.3185	0.2679	0.087*	
C19	0.1897 (3)	1.2648 (3)	0.32130 (15)	0.0641 (10)	
H19	0.2291	1.2845	0.2952	0.077*	
C20	0.2200 (2)	1.2156 (2)	0.37696 (13)	0.0434 (7)	
C21	0.3054 (2)	1.1964 (2)	0.39396 (14)	0.0457 (7)	
H21	0.3440	1.2136	0.3665	0.055*	
C22	0.33685 (18)	1.15312 (18)	0.44937 (14)	0.0408 (7)	
C23	0.4252 (2)	1.1359 (2)	0.46688 (17)	0.0510 (8)	

H23	0.4636	1.1531	0.4393	0.061*
C24	0.4557 (2)	1.0955 (2)	0.5221 (2)	0.0609 (10)
H24	0.5152	1.0864	0.5337	0.073*
C25	0.3994 (2)	1.0672 (2)	0.56234 (18)	0.0578 (9)
H25	0.4211	1.0381	0.6008	0.069*
C26	0.3150 (2)	1.08074 (19)	0.54724 (15)	0.0459 (7)
H26	0.2783	1.0599	0.5750	0.055*
C27	0.27920 (17)	1.12572 (17)	0.49037 (13)	0.0359 (6)
C28	0.40626 (19)	0.5551 (3)	0.29117 (17)	0.0573 (9)
H28A	0.3965	0.6131	0.2689	0.086*
H28B	0.4675	0.5435	0.3008	0.086*
H28C	0.3821	0.5578	0.3303	0.086*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0279 (3)	0.0485 (4)	0.0533 (5)	0.0072 (3)	0.0005 (3)	-0.0139 (3)
S2	0.0340 (4)	0.0417 (4)	0.0479 (4)	0.0033 (3)	0.0030 (3)	-0.0147 (3)
S3	0.0349 (4)	0.0816 (7)	0.1068 (8)	0.0138 (4)	0.0091 (5)	-0.0313 (6)
S4	0.0565 (7)	0.0525 (8)	0.0425 (9)	0.0104 (5)	0.0157 (5)	-0.0084 (7)
C29	0.121 (5)	0.041 (3)	0.106 (4)	-0.009 (3)	0.061 (4)	-0.033 (3)
S4A	0.108 (5)	0.083 (6)	0.070 (5)	0.044 (4)	0.023 (3)	-0.014 (4)
C29A	0.085 (16)	0.069 (14)	0.067 (16)	0.022 (11)	0.020 (13)	0.023 (13)
N1	0.0426 (12)	0.0321 (11)	0.0351 (12)	-0.0097 (10)	0.0138 (10)	-0.0043 (9)
N2	0.082 (2)	0.0406 (14)	0.0353 (14)	-0.0204 (13)	0.0083 (13)	-0.0019 (11)
N3	0.084 (2)	0.0374 (13)	0.0388 (14)	-0.0203 (13)	0.0053 (13)	0.0008 (11)
C1	0.0282 (12)	0.0273 (12)	0.0395 (14)	-0.0009 (10)	0.0040 (11)	-0.0022 (10)
C2	0.0343 (14)	0.0473 (17)	0.066 (2)	0.0055 (13)	0.0109 (14)	-0.0130 (15)
C3	0.0421 (16)	0.0500 (18)	0.063 (2)	0.0038 (14)	0.0131 (15)	-0.0183 (15)
C4	0.0271 (12)	0.0248 (11)	0.0347 (13)	-0.0013 (9)	0.0046 (10)	0.0015 (10)
C5	0.0296 (12)	0.0255 (12)	0.0349 (13)	-0.0025 (10)	0.0080 (10)	-0.0010 (10)
C6	0.0341 (13)	0.0277 (12)	0.0413 (15)	0.0017 (10)	0.0068 (11)	0.0019 (11)
C7	0.0363 (14)	0.0360 (14)	0.0371 (14)	-0.0040 (11)	0.0036 (11)	0.0009 (11)
C8	0.0383 (14)	0.0285 (12)	0.0371 (14)	-0.0082 (11)	0.0127 (11)	-0.0016 (11)
C9	0.0363 (13)	0.0282 (12)	0.0413 (15)	0.0027 (10)	0.0127 (11)	-0.0002 (11)
C10	0.0301 (12)	0.0320 (13)	0.0348 (13)	0.0002 (10)	0.0085 (10)	-0.0002 (11)
C11	0.0380 (14)	0.0347 (13)	0.0371 (15)	-0.0084 (11)	0.0115 (11)	-0.0029 (11)
C12	0.0422 (15)	0.0350 (14)	0.0353 (14)	-0.0058 (12)	0.0095 (12)	-0.0019 (11)
C13	0.0443 (15)	0.0311 (13)	0.0447 (16)	-0.0080 (12)	0.0166 (13)	-0.0090 (12)
C14	0.0378 (13)	0.0269 (12)	0.0377 (14)	-0.0062 (10)	0.0104 (11)	-0.0053 (11)
C15	0.0425 (15)	0.0304 (13)	0.0417 (15)	-0.0071 (11)	0.0057 (12)	-0.0076 (11)
C16	0.0450 (17)	0.0441 (17)	0.060 (2)	-0.0063 (13)	-0.0006 (15)	-0.0017 (15)
C17	0.061 (2)	0.067 (2)	0.073 (3)	-0.0036 (19)	-0.023 (2)	0.003 (2)
C18	0.086 (3)	0.078 (3)	0.046 (2)	-0.015 (2)	-0.014 (2)	0.0145 (19)
C19	0.086 (3)	0.065 (2)	0.0386 (18)	-0.023 (2)	0.0020 (17)	0.0035 (16)
C20	0.0565 (18)	0.0392 (15)	0.0349 (15)	-0.0153 (13)	0.0080 (13)	-0.0058 (12)
C21	0.0510 (17)	0.0458 (16)	0.0440 (16)	-0.0165 (14)	0.0194 (14)	-0.0092 (13)
C22	0.0410 (15)	0.0325 (13)	0.0512 (17)	-0.0075 (12)	0.0143 (13)	-0.0109 (13)
C23	0.0403 (16)	0.0421 (16)	0.074 (2)	-0.0066 (13)	0.0198 (16)	-0.0111 (16)
C24	0.0382 (16)	0.0437 (18)	0.099 (3)	0.0002 (14)	0.0030 (18)	-0.0094 (19)

C25	0.0546 (19)	0.0392 (16)	0.074 (2)	0.0006 (15)	-0.0084 (17)	0.0083 (16)
C26	0.0473 (16)	0.0346 (14)	0.0541 (18)	-0.0089 (13)	0.0024 (14)	0.0052 (13)
C27	0.0383 (14)	0.0261 (12)	0.0444 (15)	-0.0062 (11)	0.0096 (12)	-0.0051 (11)
C28	0.0326 (15)	0.079 (2)	0.060 (2)	0.0004 (15)	0.0090 (14)	-0.0008 (18)

Geometric parameters (Å, °)

S1—C2	1.757 (3)	C10—H10	0.9500
S1—C1	1.766 (3)	C11—C12	1.358 (4)
S2—C3	1.747 (3)	C12—H12	0.9500
S2—C1	1.751 (3)	C13—C14	1.508 (4)
S3—C2	1.745 (3)	C13—H13A	0.9900
S3—C28	1.800 (4)	C13—H13B	0.9900
S4—C3	1.760 (3)	C14—C27	1.397 (4)
S4—C29	1.817 (5)	C14—C15	1.410 (4)
C29—H29A	0.9800	C15—C16	1.428 (4)
C29—H29B	0.9800	C15—C20	1.434 (4)
C29—H29C	0.9800	C16—C17	1.367 (5)
S4A—C3	1.827 (7)	C16—H16	0.9500
S4A—C29A	1.829 (16)	C17—C18	1.396 (6)
C29A—H29D	0.9800	C17—H17	0.9500
C29A—H29E	0.9800	C18—C19	1.345 (6)
C29A—H29F	0.9800	C18—H18	0.9500
N1—N2	1.332 (3)	C19—C20	1.421 (5)
N1—C12	1.339 (3)	C19—H19	0.9500
N1—C13	1.461 (3)	C20—C21	1.382 (4)
N2—N3	1.320 (3)	C21—C22	1.379 (4)
N3—C11	1.362 (4)	C21—H21	0.9500
C1—C4	1.343 (3)	C22—C23	1.421 (4)
C2—C3	1.347 (5)	C22—C27	1.427 (4)
C4—C5	1.472 (3)	C23—C24	1.353 (5)
C4—C4 ⁱ	1.503 (5)	C23—H23	0.9500
C5—C6	1.399 (4)	C24—C25	1.403 (5)
C5—C10	1.406 (3)	C24—H24	0.9500
C6—C7	1.377 (4)	C25—C26	1.346 (4)
C6—H6	0.9500	C25—H25	0.9500
C7—C8	1.391 (4)	C26—C27	1.433 (4)
C7—H7	0.9500	C26—H26	0.9500
C8—C9	1.389 (4)	C28—H28A	0.9800
C8—C11	1.462 (4)	C28—H28B	0.9800
C9—C10	1.381 (4)	C28—H28C	0.9800
C9—H9	0.9500		
C2—S1—C1	95.96 (14)	N1—C13—C14	109.6 (2)
C3—S2—C1	96.79 (14)	N1—C13—H13A	109.8
C2—S3—C28	102.65 (15)	C14—C13—H13A	109.8
C3—S4—C29	99.2 (2)	N1—C13—H13B	109.8
C3—S4A—C29A	86.5 (9)	C14—C13—H13B	109.8
S4A—C29A—H29D	109.5	H13A—C13—H13B	108.2
S4A—C29A—H29E	109.5	C27—C14—C15	120.5 (2)

H29D—C29A—H29E	109.5	C27—C14—C13	118.8 (3)
S4A—C29A—H29F	109.5	C15—C14—C13	120.6 (3)
H29D—C29A—H29F	109.5	C14—C15—C16	123.9 (3)
H29E—C29A—H29F	109.5	C14—C15—C20	118.9 (3)
N2—N1—C12	110.8 (2)	C16—C15—C20	117.2 (3)
N2—N1—C13	121.2 (2)	C17—C16—C15	120.2 (3)
C12—N1—C13	127.6 (2)	C17—C16—H16	119.9
N3—N2—N1	107.3 (2)	C15—C16—H16	119.9
N2—N3—C11	108.4 (2)	C16—C17—C18	122.3 (4)
C4—C1—S2	120.55 (19)	C16—C17—H17	118.9
C4—C1—S1	126.5 (2)	C18—C17—H17	118.9
S2—C1—S1	112.87 (14)	C19—C18—C17	119.3 (3)
C3—C2—S3	123.9 (2)	C19—C18—H18	120.4
C3—C2—S1	117.1 (2)	C17—C18—H18	120.4
S3—C2—S1	118.77 (19)	C18—C19—C20	121.6 (4)
C2—C3—S2	116.6 (2)	C18—C19—H19	119.2
C2—C3—S4	126.9 (2)	C20—C19—H19	119.2
S2—C3—S4	116.51 (19)	C21—C20—C19	121.4 (3)
C2—C3—S4A	113.0 (3)	C21—C20—C15	119.2 (3)
S2—C3—S4A	119.3 (3)	C19—C20—C15	119.4 (3)
C1—C4—C5	125.3 (2)	C22—C21—C20	122.4 (3)
C1—C4—C4 ⁱ	116.4 (2)	C22—C21—H21	118.8
C5—C4—C4 ⁱ	118.0 (2)	C20—C21—H21	118.8
C6—C5—C10	116.7 (2)	C21—C22—C23	121.6 (3)
C6—C5—C4	124.7 (2)	C21—C22—C27	119.1 (3)
C10—C5—C4	118.6 (2)	C23—C22—C27	119.2 (3)
C7—C6—C5	122.1 (2)	C24—C23—C22	121.2 (3)
C7—C6—H6	118.9	C24—C23—H23	119.4
C5—C6—H6	118.9	C22—C23—H23	119.4
C6—C7—C8	120.3 (3)	C23—C24—C25	119.9 (3)
C6—C7—H7	119.9	C23—C24—H24	120.0
C8—C7—H7	119.9	C25—C24—H24	120.0
C9—C8—C7	118.5 (2)	C26—C25—C24	120.9 (3)
C9—C8—C11	119.0 (2)	C26—C25—H25	119.5
C7—C8—C11	122.3 (3)	C24—C25—H25	119.5
C10—C9—C8	121.0 (2)	C25—C26—C27	121.7 (3)
C10—C9—H9	119.5	C25—C26—H26	119.1
C8—C9—H9	119.5	C27—C26—H26	119.1
C9—C10—C5	121.1 (2)	C14—C27—C22	119.8 (3)
C9—C10—H10	119.5	C14—C27—C26	123.3 (3)
C5—C10—H10	119.5	C22—C27—C26	116.9 (3)
C12—C11—N3	107.9 (2)	S3—C28—H28A	109.5
C12—C11—C8	126.8 (3)	S3—C28—H28B	109.5
N3—C11—C8	125.1 (2)	H28A—C28—H28B	109.5
N1—C12—C11	105.5 (2)	S3—C28—H28C	109.5
N1—C12—H12	127.2	H28A—C28—H28C	109.5
C11—C12—H12	127.2	H28B—C28—H28C	109.5
C12—N1—N2—N3	0.3 (3)	C9—C8—C11—C12	-36.6 (4)

C13—N1—N2—N3	-172.9 (3)	C7—C8—C11—C12	139.4 (3)
N1—N2—N3—C11	-0.3 (4)	C9—C8—C11—N3	147.6 (3)
C3—S2—C1—C4	169.8 (2)	C7—C8—C11—N3	-36.3 (4)
C3—S2—C1—S1	-7.97 (19)	N2—N1—C12—C11	-0.2 (3)
C2—S1—C1—C4	-170.3 (3)	C13—N1—C12—C11	172.5 (3)
C2—S1—C1—S2	7.28 (18)	N3—C11—C12—N1	0.0 (3)
C28—S3—C2—C3	155.2 (3)	C8—C11—C12—N1	-176.4 (3)
C28—S3—C2—S1	-30.0 (3)	N2—N1—C13—C14	121.3 (3)
C1—S1—C2—C3	-3.6 (3)	C12—N1—C13—C14	-50.6 (4)
C1—S1—C2—S3	-178.7 (2)	N1—C13—C14—C27	-79.6 (3)
S3—C2—C3—S2	173.3 (2)	N1—C13—C14—C15	97.9 (3)
S1—C2—C3—S2	-1.5 (4)	C27—C14—C15—C16	178.1 (3)
S3—C2—C3—S4	-7.9 (5)	C13—C14—C15—C16	0.7 (4)
S1—C2—C3—S4	177.2 (2)	C27—C14—C15—C20	-1.7 (4)
S3—C2—C3—S4A	29.4 (5)	C13—C14—C15—C20	-179.1 (2)
S1—C2—C3—S4A	-145.4 (4)	C14—C15—C16—C17	177.0 (3)
C1—S2—C3—C2	5.9 (3)	C20—C15—C16—C17	-3.2 (4)
C1—S2—C3—S4	-173.1 (2)	C15—C16—C17—C18	-0.2 (6)
C1—S2—C3—S4A	147.4 (4)	C16—C17—C18—C19	2.9 (6)
C29—S4—C3—C2	91.4 (4)	C17—C18—C19—C20	-2.0 (6)
C29—S4—C3—S2	-89.8 (3)	C18—C19—C20—C21	-179.8 (3)
C29—S4—C3—S4A	14.2 (5)	C18—C19—C20—C15	-1.5 (5)
C29A—S4A—C3—C2	-130.5 (9)	C14—C15—C20—C21	2.1 (4)
C29A—S4A—C3—S2	86.7 (9)	C16—C15—C20—C21	-177.7 (3)
C29A—S4A—C3—S4	-8.5 (8)	C14—C15—C20—C19	-176.2 (3)
S2—C1—C4—C5	-176.68 (19)	C16—C15—C20—C19	4.0 (4)
S1—C1—C4—C5	0.7 (4)	C19—C20—C21—C22	176.5 (3)
S2—C1—C4—C4 ⁱ	-2.8 (3)	C15—C20—C21—C22	-1.8 (4)
S1—C1—C4—C4 ⁱ	174.61 (18)	C20—C21—C22—C23	-178.4 (3)
C1—C4—C5—C6	-27.2 (4)	C20—C21—C22—C27	1.0 (4)
C4 ⁱ —C4—C5—C6	159.0 (2)	C21—C22—C23—C24	178.2 (3)
C1—C4—C5—C10	150.5 (3)	C27—C22—C23—C24	-1.2 (4)
C4 ⁱ —C4—C5—C10	-23.3 (3)	C22—C23—C24—C25	2.1 (5)
C10—C5—C6—C7	-4.2 (4)	C23—C24—C25—C26	-0.9 (5)
C4—C5—C6—C7	173.5 (2)	C24—C25—C26—C27	-1.1 (5)
C5—C6—C7—C8	-0.8 (4)	C15—C14—C27—C22	1.0 (4)
C6—C7—C8—C9	4.3 (4)	C13—C14—C27—C22	178.4 (2)
C6—C7—C8—C11	-171.8 (2)	C15—C14—C27—C26	-179.4 (2)
C7—C8—C9—C10	-2.6 (4)	C13—C14—C27—C26	-2.0 (4)
C11—C8—C9—C10	173.6 (2)	C21—C22—C27—C14	-0.6 (4)
C8—C9—C10—C5	-2.5 (4)	C23—C22—C27—C14	178.8 (2)
C6—C5—C10—C9	5.8 (4)	C21—C22—C27—C26	179.8 (3)
C4—C5—C10—C9	-172.0 (2)	C23—C22—C27—C26	-0.8 (4)
N2—N3—C11—C12	0.2 (4)	C25—C26—C27—C14	-177.7 (3)
N2—N3—C11—C8	176.6 (3)	C25—C26—C27—C22	1.9 (4)

Symmetry code: (i) $-x, y, -z+1/2$.

Hydrogen-bond geometry (Å, °)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C28—H28C \cdots N2 ⁱⁱ	0.98	2.55	3.495 (5)	161
C25—H25 \cdots S2 ⁱⁱⁱ	0.95	3.00	3.784 (4)	141

Symmetry codes: (ii) $-x+1/2, -y+3/2, -z+1$; (iii) $x+1/2, -y+3/2, z+1/2$.